	レビン	
Name:	KU /	Block:

# Assignment #2 – Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
	The ionic charge is given after the first element (metallic) in Roman
Numerals.	

1. copper (II) oxide CUO	11. manganese (III) oxide Mv 203		
2. mercury (I) oxide HQ20	12. vanadium (II) bromide VB(z		
3. gold (III) chloride _AuCl3	13. niobium(V) oxide Nb205		
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide		
5. bismuth (V) oxide	15. titanium (III) nitride TIN		
6. terbium (III) oxide	16. iron (II) oxide <u>Fe.O</u>		
7. uranium (VI) oxide \( \lambda \ \O \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	17. cobalt (II) phosphide C03P2		
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide		
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2		
10. arsenic (V) sulphideAS_2S_5	20. copper (II) bromide CUPY2		
B. Name the compound using the Roman Numeral Method and balance the ionic charges.			
1. SnCl Tin (IV) Chloride	9. Pdf. Palladium (IV) Fluoride		
2. BIBIS BISMINTO STOMICE	10. OszOs OSMIUM (III) Oxide		
3. POO2 ROLONIUM (IV) OXICE	11. MOBEZ MOLYBORENIUM (II) Bramide		
4. Pbl2 100d(II) Iodide	12. VCIs Vanadium (V) Chloride		
5. HgO <u>Mercury</u> (II) Oxide	13. Mn2O3 Manganese (III) Oxide		
6. HgCI MEYCUYY (I) Chroride	14. COO COLONY (II) Oxide		
7. Auzos Gold (III) Oxide	15. Np2O3 NEDTUNIUM (II) OXICE		
8. FeClo Iron (II) Chloride	16. V2O6 Vanadium (V) Oxide		

## **Answers For Writing Formulas From Names**

Morris Hein, Scott Pattison, Susan Arena, Leo R. Best

#### **Answers For Writing Formulas From Names:**

A Textbook of Organic Chemistry, 22e Arun Bahl & B S Bahl, 2019 With an increased focus on fundamentals this new edition of A Textbook of Organic Chemistry continues to present the time tested functional group approach to the subject This examination oriented book breaks the intricacies of Organic Chemistry into easy to understand steps which gives the student the necessary foundation to build upon learn and understand Organic Chemistry in a way that is efficient as well as Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers Chemistry: Concepts and Problems Clifford C. Houk, Richard Post, 1996-03-09 CHEMISTRY SECOND and students EDITION The fast easy way to master the fundamentals of chemistry Have you ever wondered about the differences between liquids gases and solids Or what actually happens when something burns What exactly is a solution An acid A base This is chemistry the composition and structure of substances composing all matter and how they can be transformed Whether you are studying chemistry forthe first time on your own want to refresh your memory for a test or need a little help for a course this concise interactive guidegives you a fresh approach to this fascinating subject This fullyup to date edition of Chemistry Concepts and Problems Has been tested rewritten and retested to ensure that you canteach yourself all about chemistry Requires no prerequisites Lets you work at your own pace with a helpful question and answerformat Lists objectives for each chapter you can skip ahead or findextra help if you need it Reinforces what you learn with chapter self tests Chemical A Textbook of Organic Chemistry, 23e Arun Bahl & B S Bahl, A Textbook of formulas and names Virginia P. Powell, 1965

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