

Data Report Sheet: Analysis of an Unknown Chloride

Unknown # 32

Molarity of Standard $\text{Ag}(\text{NO}_3)_3$ 0.1482 Moles/liter

	Trial I	Trial II	Trail III
Mass of Unknown	0.3991 g	0.3991 g	0.3991 g
Initial Burette Reading	49.3	29.5	37.6 ML
Final Burette Reading	29.5	9.8	17.8ML
Vol. $\text{Ag}(\text{NO}_3)$ Added	?	?	?
Moles $\text{Ag}(\text{NO}_3)$	?	?	?
Moles Cl^-	?	?	?
Mass Cl^-	?	?	?
% Cl^- in Unknown	?	?	?

Average % Cl^- in unknown ? %

Analysis Of An Unknown Chloride Answers

J Spring



Analysis Of An Unknown Chloride Answers:

Environmental Sampling and Analysis Maria Csuros,1997-03-24 This manual covers the latest laboratory techniques state of the art instrumentation laboratory safety and quality assurance and quality control requirements In addition to complete coverage of laboratory techniques it also provides an introduction to the inorganic nonmetallic constituents in environmental samples their chemistry and their control by regulations and standards Environmental Sampling and Analysis Laboratory Manual is perfect for college and graduate students learning laboratory practices as well as consultants and regulators who make evaluations and quality control decisions Anyone performing laboratory procedures in an environmental lab will appreciate this unique and valuable text **Plating** ,1950 Cambridge IGCSE® Chemistry Practical Teacher's Guide with CD-ROM Michael Strachan,2016-06-23 This edition of our successful series to support the Cambridge IGCSE Chemistry syllabus 0620 is fully updated for the revised syllabus from first examination from 2016 The Cambridge IGCSE Chemistry Practical Teacher s Guide complements the Practical Workbook helping teachers to include more practical work in lessons Specific support is provided for each of the carefully designed investigations to save teachers time The Teacher s Guide contains advice about planning investigations guidance about safety considerations differentiated learning suggestions to support students who might be struggling and to stretch the students who are most able as well as answers to all the questions in the Workbook The Teacher s Guide also includes a CD ROM containing model data to be used in instances when an investigation cannot be carried out *Analytical Chemistry* Gary D. Christian,Purnendu K. Dasgupta,Kevin A. Schug,2013-10-07 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry **First Course in Quantitative Analysis** Floyd Hamilton Fish,1931 **Introduction to Drug Analysis** Robert Francis Raffauf,Victor D. Warner,1978 *IIT JEE Chemistry Exam Leaders Expert*, **Clinical Chemistry** ,1959 **English Mechanic and Mirror of Science and Art** ,1893

Quantitative Analysis William Rieman,Jacob David Neuss,Barnet Naiman,1951 Common apparatus and operations The balance Volumetric apparatus Fundamentals of volumetric analysis Volumetric determination of chloride ion by Mohr s method Potentiometric measurements Ionization of salts acids and bases Acidimetry and alkalimetry Manual of Qualitative Chemical Analysis C. Remigius Fresenius,Horace Lemuel Wells,1897 Student Self-study Guide Donald W. Shive,1979 Manual of qualitative chemical analysis Karl Remigius Fresenius,1897 **General Chemistry** Raymond Chang,1986 **General Chemistry** Frank Brescia,1988 **Instrumental Methods of Chemical Analysis** Galen Wood

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