Types of Chemical Reactions Lab Answer Key

Chemical Reaction #1

Reactants (write in words)	hydrochloric acid and magnesium	
Observations	accept most responses	
Type of Reaction	single replacement	
Predict Products (write in words)	magnesium chloride and hydrogen gas	
Equation (write in symbols)	$2HCl_{log} + Mg_{log} \rightarrow MgCl_{log} + H_{log}$	

Chemical Reaction #2

Reactants (write in words)	copper metal and oxygen gas
Observations	accept most responses
Type of Reaction	synthesis
Predict Products (write in words)	copper oxide
Equation (write in symbols)	2Cu ₀₀ + O ₂₀₀ → 2CuO ₀₀

Chemical Reaction #3

Reactants (write in words)	methane gas and oxygen gas	
Observations	accept most responses	
Type of Reaction	combustion	
Predict Products (write in words)	carbon dioxide and water	
Equation (write in symbols)	$CH_{tigs} + 2O_{2igs} \rightarrow CO_{2igs} + 2H_2O_{ilis}$	

Chemical Reaction #4

Reactants (write in words)	hydrogen peroxide
Observations	accept most responses
Type of Reaction	decomposition
Predict Products (write in words)	water and oxygen gas
Equation (write in symbols)	$2H_2O_{20} \rightarrow 2H_2O_{0} + O_{200}$

Answer Key To Types Of Chemical Reactions Lab 14

J Spring

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