INTRODUCTION

LABORATORY SIMULATION

ALTERNA V	
ACCOUNT OF THE PARTY OF THE PAR	
Market Street	

ab Data

Verify your calculation. In your set-up, do your units cancel?

Mass of magnesium (g)	.015
Moles of magnesium (mol)	.00062
Temperature of water (°C)	23.0
Temperature of water (K)	296.15
Vapor pressure of water (mmHg)	21.10
Barometric pressure (mmHg)	778.0
servations	

Observ

I observe that there are bubbles that are visible.

Volume of hydrogen gas collected (mL)	15.00
Volume of hydrogen gas collected (L)	
Pressure of hydrogen gas (mmHg)	0.01500
Pressure of hydrogen gas (atm)	756.9
Moles of hydrogen gas produced (mol)	0.9959
Ideal gas constant (L-atm-mol ⁻¹ -K ⁻¹)	.00690
	0.080



























ES



Answers Escience Lab Ideal Gas Law

Edgar Buckingham

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