

**Note:** In communicative exercises where several answers are possible, this answer key contains some examples of correct answers, not all possible answers. Any valid answer in this type of exercise should be considered acceptable.

**Exercise 1** Answers will vary.

**Exercise 2**

1. I can't get over how much I enjoy it. / I can't get enough of it. 2. It's a pain in the neck. / I've had about enough of it. 3. It's more trouble than it's worth. 4. It's a pain in the neck. / I've had about enough of it. / It's more trouble than it's worth. 5. I can't get over how much I enjoy it. / I can't get enough of it.

**Exercise 3**

a. to call b. eating c. to make d. meeting e. to tell  
1. b 2. e 3. a 4. d 5. c

**Exercise 4** Answers will vary.

**Exercise 5**

1. Remember to buy coffee. / Don't forget to buy coffee.  
2. Remember to wish your wife happy anniversary. / Don't forget to wish your wife happy anniversary. 3. Stop working so much. 4. Remember to turn off your cell phone. / Don't forget to turn off your cell phone.

**Exercise 6** 1. c 2. a 3. d 4. b

**Exercise 7** Answers will vary.

**Exercise 8** Answers will vary.

**Exercise 9**

Answers will vary. Following are examples of appropriate responses:

A: Have you had a chance to meet the new employee?

B: No, I haven't. I wonder what she's like.

A: Well, everyone says she's a pain in the neck.

B: Yeah, but you can't believe everything you hear. She could just be shy.

**Exercise 10**

1. P 2. P 3. O 4. O 5. P 6. P  
1. O 2. P 3. O 4. P 5. P 6. O

**Exercise 11**

1. to look 2. expecting 3. to see 4. believing 5. to avoid 6. to try

**Exercise 12** Answers will vary.

**Exercise 13**

Answers will vary but may include the following: 1. He was diagnosed with cancer. 2. He was optimistic. He found treatments, and he trained. He prepared for his future. 3. Yes. He survived cancer and won the Tour de France seven times. 4. Answers will vary.

**Exercise 14** Answers will vary.

**Exercise 15** 1. b 2. a 3. c

**Exercise 16** Answers will vary.

**Exercise 17** Answers will vary.

## GRAMMAR BOOSTER

**Exercise A**

1. to come 2. to spend 3. to try, slowing  
4. to disappoint, advancing 5. to live, working, bringing  
6. to take, doing

**Exercise B**

Answers will vary. Following are examples of appropriate responses: 1. going on vacation next week 2. living in the city 3. to have a wonderful family 4. to meet my deadlines 5. getting spam 6. to pursue my interests 7. listening to music 8. to learn to fly a plane 9. watching ads on TV 10. to get a better job

**Exercise C** Answers will vary.

# Answer Summit 1a Unit 3

**Dr. Amar Bhutani**



### **Answer Summit 1a Unit 3:**

*English in Mind Levels 1A and 1B Combo Teacher's Resource Book* Brian Hart, 2011-02-10 This second edition updates a course which has proven to be a perfect fit for classes the world over The Teacher's Resource Book contains the content for Combos 1A and 1B in one volume All unit numbers and page references correspond to the Combos It contains extra photocopiable grammar and communication activities and full pages of teaching tips and ideas specially written by methodology expert Mario Rinvolucri A Combo Testmaker Audio CD CD ROM which allows teachers to create and edit their own tests is also available separately as is Classware for the full edition which integrates the Student's Book class audio and video *Nelson English - Book 3 Teacher's Guide* John Jackman, Wendy Wren, 2000 Each pair of units is supported by the Teachers Guide The Teachers Guide fully supports the programme giving all the guidance you need to help you pupils work through the two pupil books In the Teachers Guide there are suggestions for homework and independent study

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction  $2\text{Ag}^+ \text{aq} + \text{CrO}_4^{2-} \text{aq} \rightarrow \text{Ag}_2\text{CrO}_4 \text{s}$

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is  $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation  $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$  the spectator ions are  $\text{Cl}^-$  and  $\text{Na}^+$  net ionic equation  $\text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$  here is a video on spectator ions *chapter 8 04 ionic equations chemistry libretexts* - Mar 06 2023

web the six  $\text{NO}_3^- \text{aq}$  ions and the six  $\text{Na}^+ \text{aq}$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3(\text{aq}) + \text{KCl}(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{KNO}_3(\text{aq})$  total ionic  $\text{Ag}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s}) + \text{K}^+(\text{aq}) + \text{NO}_3^-(\text{aq})$  net ionic  $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+(\text{aq}) + \text{OH}^-(\text{aq}) \rightarrow \text{H}_2\text{O}(\text{l})$  the spectator ions in  $\text{KOH}(\text{aq})$  and  $\text{HNO}_3(\text{aq})$  are  $\text{K}^+(\text{aq})$  and  $\text{NO}_3^-(\text{aq})$  how to identify and list the

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web the  $\text{Ag}^+(\text{aq})$  and  $\text{Cl}^-(\text{aq})$  ions become  $\text{AgCl}(\text{s})$  but the  $\text{Na}^+(\text{aq})$  ions and the  $\text{NO}_3^-(\text{aq})$  ions stay as  $\text{Na}^+(\text{aq})$  ions and  $\text{NO}_3^-(\text{aq})$  ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be  $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{Cu}^{2+} \text{aq} + 2\text{Cl}^- \text{aq}$  these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

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