

## Quick Lab

**Purpose** To use a graph to identify period and group trends

### Materials

- graph paper
- pencil



## Periodic Trends in Ionic Radii

### Procedure

Use the data presented in Figure 6.23 to plot ionic radius versus atomic number.

### Analyze and Conclude

- 1. Compare** How does the size change when an atom forms a cation and when an atom forms an anion?
- 2. Describe** How do the ionic radii vary within a group of metals? How do they vary within a group of nonmetals?
- 3. Describe** What is the shape of a portion of the graph that corresponds to one period?
- 4. Compare and Contrast** Is the trend across a period similar or different for Periods 2, 3, 4, and 5?
- 5. Explain** Propose explanations for the trends you have described for ionic radii within groups and across periods.

**Figure 6.23 Atomic and Ionic Radii**  
Atomic and ionic radii are an indication of the relative size of atoms and ions. The data listed are reported in picometers (pm).



# Answers Quick Lab Periodic Trends In Ionic Radii

**Mark Steyvers**



## **Answers Quick Lab Periodic Trends In Ionic Radii:**

## **Answers Quick Lab Periodic Trends In Ionic Radii** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the energy of words has been more evident than ever. They have the ability to inspire, provoke, and ignite change. Such could be the essence of the book **Answers Quick Lab Periodic Trends In Ionic Radii**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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