

Molecular Polarity Lab

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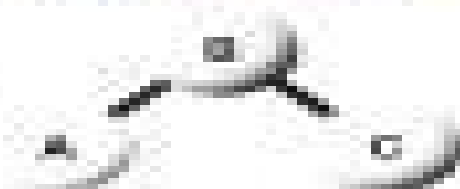
Circle the best correct choice. Check the right responses (the checkboxes).

Part A: Three atom investigations

1. Lightly keep the box for bond dipole moments and check responses for about three substances and the bond dipole moment.
 - a. Check the electronegativity of A and B and compare and discuss the electronegativity of atom B (drawn in the arrow, partial charges and bond structure).
 1. Write the electronegativity of atom B (use last digit) _____
 2. partial charges _____
 3. bond structure _____
2. Check partial dipole moment for a series of three substances (electronegativity of B and answer the results). Fill in the following information:
 - a. The polarity arrow points to the _____ electronegativity.
 - b. The partial positive charge is always on the _____ electronegative atom.
 - c. The larger the electronegativity difference the more _____ the bond structure.

Part B: Three atom investigations

1. Check the central atom electronegativity and the electronegativity of A and C and compare. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color, describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbols on the Lewis dot diagram.
 - d. Circle the electronegative. Add other partial responses for molecules - for example:
2. Check the central atom electronegativity of A and B and compare. Circle to show the bond dipole, molecular dipole and partial charges.
 - a. Draw the bond dipole on the Lewis dot diagram.
 - b. In a different color, describe partial molecular dipole on the diagram.
 - c. Draw the partial charge symbols on the Lewis dot diagram.
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