

Experiment 17 Report Sheet

Antacid Analysis

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. Dissolving the Antacid

	Trial 1	Trial 2	Trial 1	Trial 2
1. Mass of flask (g)	128.457	97.342	115.395	111.220
2. Mass of flask + antacid sample (g)	128.661	97.542	115.606	111.440
3. Mass (or tared mass) of antacid sample (g)	.204	.200	0.211	0.217
4. Total volume of HCl added (mL)	25	25	25	25
5. Molar concentration of HCl (mol/L)	.09687		0.9687	

B. Analyzing the Antacid Sample

1. Molar concentration of NaOH (mol/L)	.09361		0.9361	
2. Buret reading, initial (mL)	1.5	16.7	9.7	11.6
3. Buret reading, final (mL)	16.7	28.4	18.2	21.7
4. Volume of NaOH (mL)	15.2	12.3	8.5	10.1

C. Calculations

1. Moles of HCl added, total (mol)				
2. Moles of NaOH added (mol)				
3. Moles of base in antacid sample (mol)				
4. $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)		*		*
5. Average $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)				

*Show calculation(s) for Trial(s) 1 on the next page.

Analysis Of Antacids Lab Report Answers

Karen C. Timberlake



Analysis Of Antacids Lab Report Answers:

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Science Web Peter Horsfall, 2001-08-15 The Science Web series provides resources that cover National Curriculum Key Stage 3 science and the approaches outlined in the QCA scheme of Work for Science This enquiry pack includes student and teacher materials back to back for easy reference and management guidance and notes for technicians materials suitable for individual and group work and suggestions on the use of ICT to support the development of enquiry skills **Plumb's**

Veterinary Drug Handbook Donald C. Plumb, 2018-02-21 Plumb's Veterinary Drug Handbook Ninth Edition updates the most complete detailed and trusted source of drug information relevant to veterinary medicine Provides a fully updated edition of the classic veterinary drug handbook with carefully curated dosages per indication for clear guidance on selecting a dose Features 16 new drugs Offers an authoritative complete reference for detailed information about animal medication Designed to be used every day in the fast paced veterinary setting Includes dosages for a wide range of species including dogs cats exotic animals and farm animals Abstracts of the XI International Congress of Gastroenterology, 1980

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Abstracts of the ... International Congress of Gastroenterology*, A.S.N.E.M.G.E., 1980 **Study Guide for Kinn's Medical Assisting Fundamentals E-Book** Brigitte

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Cumulated Index Medicus, 1993

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Carbon-14 emits beta radiation and decays with a half-life ($t_{1/2}$) of 5730 years. Assume you start with a mass of 2.00 10 12 g
of carbon-14. a. How long is ... ECON101 - Ch.25 Section Review Answers For the electronic transition from $n = 3$ to $n = 5$ in
the hydrogen atom. a) Calculate the energy. b) Calculate the wavelength (in nm). Chapter 25 Nuclear Chemistry 25.2 Nuclear
Transformations Sep 5, 2017 — Nuclear Chemistry Targets: 1.I CAN Utilize appropriate scientific vocabulary to explain
scientific concepts. 2.I CAN Distinguish between fission ... Matter and Change • Chapter 25 When a radioactive nucleus gives
off a gamma ray, its atomic number increases by. 12. The three types of radiation were first identified by Ernest Rutherford.
Nuclear Chemistry - Lake Central High School Jul 12, 2015 — What is the change in atomic number after the alpha decay?It
decreases by 2.b. ... answer the following questions.Nuclear ... 25.2 Nuclear Transformations | Lecture
notes Chemistry These nuclei decay by turning a neutron into a pro- ton to emit a beta particle (an electron) from the
nucleus. This process is known as beta emission. It ... 60 s - 1 min SECTION 25.2 NUCLEAR TRANSFORMATIONS. 1. Write a
nuclear equation for the following radioactive processes. a. alpha decay of francium-208 $^{208}\text{Fr} \rightarrow \text{b} \dots$