

Lab 6: Nitro Reduction

Pre-lab Assignment: – Research online resources and writing Lab Report before coming.

Start at the beginning of the 1st lab session:

1. Complete the following in your lab notebook:

- Reading the Labster Lecture
- Completing the reaction information at the top of the report (1st page)
- Answering Questions on the report your notebook, check your lab notebook
- On the 1st page when the synthesis is complete (all pages), make a copy of the entire report and bring it to the 2nd lab session as part of your lab notebook. Also bring your lab notebook to the 2nd lab session. Complete the report in the notebook, with your notebook information already included for you.

Start at the start of the 2nd lab session:

1. The completed labster report (all pages) will be accepted by the instructor as part of the labster assignment. Handwritten notes (in your notebook) will be accepted as part of the labster assignment.
2. Complete the following in your lab notebook:
 - a. **Pre-lab:** Write your own pre-lab for this lab. The pre-lab should include a brief description of the reaction and the reaction conditions. Also include your pre-lab.
 - b. **Lab:** Write your lab report (all pages) that you bring to the 2nd lab session. The labster report will be accepted as part of the labster assignment.
3. Complete the labster report and pre-lab report (all pages).

INTRODUCTION

Reduction of nitro compounds with a metal catalyst is a common reaction. In this lab, you will be using a metal catalyst to reduce nitro compounds. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group.

The important reaction mechanism is the reduction of nitro compounds. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group.

Reaction 1: Reduction of Nitrobenzene to Aniline

The reaction of nitrobenzene with a metal catalyst is a common reaction. In this lab, you will be using a metal catalyst to reduce nitrobenzene to aniline. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group. The reaction is a redox reaction. The metal catalyst is used to reduce the nitro group to an amino group.

Analysis On The Nine Solutions Lab

Harry Sands Grindley



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