2009 AP" CHEMISTRY FREE-RESPONSE QUESTIONS

Answer Question 5 and Question 6. The Section II score weighting for these questions is 15 percent each.

Your responses to these questions will be graded on the basis of the accuracy and relevance of the information cited. Explanations should be clear and well organized. Examples and equations may be included in your responses where appropriate. Specific answers are preferable to broad, diffuse responses.

| Reaction | Equation | ΔH_{298}° | ΔS_{Sys}^{α} | ΔG_{Pes}^{a} |
|----------|---|---------------------------|--|--------------------------|
| x | $C(s) + H_2O(g) \rightleftharpoons CO(g) + H_2(g)$ | +131 kJ mol ⁻¹ | +134 J mol ⁻¹ K ⁻¹ | +91 kJ mol ⁻¹ |
| Y | $CO_2(g) + H_2(g) \rightleftharpoons CO(g) + H_2O(g)$ | +41 kJ mol ⁻¹ | +42 J mol ⁻¹ K ⁻¹ | +29 kJ mol ⁻¹ |
| z | 2 CO(g) | 20 | 7 | .7 |

- Answer the following questions using the information related to reactions X, Y, and Z in the table above.
 - (a) For reaction X, write the expression for the equilibrium constant, K₋.
 - (b) For reaction X, will the equilibrium constant, K_p, increase, decrease, or remain the same if the temperature rises above 298 K? Justify your answer.
 - (c) For reaction Y at 298 K, is the value of K_w greater than 1, less than 1, or equal to 1? Justify your answer.
 - (d) For reaction Y at 298 K, which is larger: the total bond energy of the reactants or the total bond energy of the products? Explain.
 - (e) Is the following statement true or false? Justify your answer.

"On the basis of the data in the table, it can be predicted that reaction Y will occur more rapidly than reaction X will occur."

- (f) Consider reaction Z at 298 K.
 - Is ΔS° for the reaction positive, negative, or zero? Justify your answer.
 - (ii) Determine the value of ΔHⁿ for the reaction.
 - (iii) A sealed glass reaction vessel contains only CO(g) and a small amount of C(s). If a reaction occurs and the temperature is held constant at 298 K, will the pressure in the reaction vessel increase, decrease, or remain the same over time? Explain.

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problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner **Evolution in Reference and Information Services** Linda S Katz, 2013-05-13 Explore ways to bring and keep your library s electronic services up to date From editor Di Su Some years ago if you were told that a library s catalog would be available on a 24 7 365 basis you d think it was just another fiction Perhaps as influential as Johannes Gutenberg s invention of movable type printing the Internet is one of the most significant happenings in the information world in modern times In addition to showing you how library services have been influenced and enhanced by the advent of the Internet Evolution in Reference and Information Services The Impact of the Internet will enable you to make the most of the new opportunities that current technologies offer This valuable book will also help you and your library avoid the pitfalls and new challenges to professional competency that come along with electronic research Evolution in Reference and Information Services gives you a review of the history of electronic reference looks at the increasing role of librarians as teachers and providers of technical help for users provides case studies and ways to evaluate electronic research methods suggests strategies for providing effective electronic services examines government Web sites explores Internet sources of health information shows you how to establish electronic services through your library s portal site looks at how to manage a library computer lab and much more SAGE Handbook of Research on Classroom Assessment James H. McMillan, 2013 The Sage Handbook of Research on Classroom Assessment provides scholars professors graduate students and other researchers and policy makers in the organizations agencies testing companies and school districts with a comprehensive source of research on all aspects of K 12 classroom assessment The handbook emphasizes theory conceptual frameworks and all varieties of research quantitative qualitative mixed methods to provide an in depth understanding of the knowledge base in each area of classroom assessment and how to conduct inquiry in the area It presents classroom assessment research to convey in depth the state of knowledge and understanding that is represented by the research with particular emphasis on how classroom assessment practices affect student achieventment and teacher behavior Editor James H McMillan and five Associate Editors bring the best thinking and analysis from leading classroom assessment researchers on the nature of the research making significant contributions to this prominent and hotly debated topic in education Questions and answers for job interview Offshore Oil & Gas Platforms Petrogav International

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startled frankly by the apparent lack of interest in the history of our field and decided to attempt to compile a history ourselves As is the case with any edited volume the contributing authors deserve credit for its positive features They uniformly made every effort asked of them and taught us much about educational psychology Any errors or omissions are our responsibility alone Job interview questions and answers for employment on Offshore Oil & Gas Rigs Petrogav International Oil & Gas Training Center, 2020-07-01 The job interview is probably the most important step you will take in your job search journey Because it s always important to be prepared to respond effectively to the questions that employers typically ask at a job interview Petrogav International has prepared this eBooks that will help you to get a job in oil and gas industry Since these questions are so common hiring managers will expect you to be able to answer them smoothly and without hesitation This eBook contains 289 questions and answers for job interview and as a BONUS web addresses to 289 video movies for a better understanding of the technological process This course covers aspects like HSE Process Mechanical Electrical and Instrumentation Control that will enable you to apply for any position in the Oil and Gas Industry to a 5 AP Environmental Science, 2014-2015 Edition Linda Williams, 2013-07-09 Get ready for your AP exam with this straightforward and easy to follow study guide updated for all the latest exam changes 5 Steps to a 5 AP Environmental Science features an effective 5 step plan to guide your preparation program and help you build the skills knowledge and test taking confidence you need to succeed This fully revised edition covers the latest course syllabus and provides model tests that reflect the latest version of the exam Inside you will find 5 Step Plan to a Perfect 5 1 Set Up Your Study Program 2 Determine Your Test Readiness 3 Develop Strategies for Success 4 Develop the Knowledge You Need to Score High 5 Build Your Test Taking Confidence 2 complete practice AP Environmental Science exams 3 separate plans to fit your study style Review material updated and geared to the most recent tests Savvy information on how tests are constructed scored and Food Chemistry in Small Bites Patricia B O'Hara, 2025-04-15 Food Chemistry in Small Bites takes readers on an used up close scientific journey through the transformation of food when meals are prepared Organized in bite size digestible units this innovative text introduces students to food s molecular makeup as well as the perception of food by the five senses Using familiar foods as examples it explores what happens to ingredients when heated cooled or treated and also considers what happens when materials that don t naturally mix are forced to do so With informative full color renderings and a hands on lab section the book encourages students to think like scientists while preparing delicious dishes Readers will formulate hypotheses as to why certain foods taste hot despite being at room temperature why milk separates into curds and whey when lemon is added and other ordinary but chemically complex phenomena This book also importantly challenges readers to think critically about the future of food in the face of a warming planet

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