

16. Given the chemical formula of an ionic compound, list at least three necessary steps to give the correct name of that compound. (If needed, use a chemical formula of a compound from the table above as an example in listing the naming steps.)

CHEMICAL BONDING POGIL

Activity 4 – Predicting the correct chemical formula for ionic compounds formed from simple anions

Objective:

- To learn how to predict the correct number of cations and anions in a simple salt.

The Model:

Table 3

Cations	Anions	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Na^+	S^{2-}	Na_2S	sodium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Exploring the Model

1. What is the charge on the zinc ion?
2. What is the charge on the nitride ion?
3. What is the charge on the chloride ion?
4. What is the charge on the ionic compound, sodium chloride?
5. What is the charge on the ionic compound, sodium sulfide?
6. How many potassium ions are present in K_3N ?
7. What does the “2” stand for in the formula for ZnCl_2 ?

Exploring the Model

8. Sodium chloride is NaCl , and zinc chloride is ZnCl_2 . Why are there more chloride ions in the zinc compound?
9. Sodium chloride is NaCl , and sodium sulfide is Na_2S . Why are there more sodium ions in the sulfide compound?

Answers Chemical Bonding Pogil

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Cooper, Thomas J. Greenbowe**



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