

# Grade 11 Chemistry

## INTERMOLECULAR FORCES & Physical Properties

### EXAM QUESTIONS PART 3



| NAME OF SUBSTANCE | FORMULA                            | BOILING POINT (°C) |
|-------------------|------------------------------------|--------------------|
| Water             | H <sub>2</sub> O                   | 100                |
| Ethanol           | CH <sub>3</sub> CH <sub>2</sub> OH | 78                 |
| Bromine           | Br <sub>2</sub>                    | 58,8               |
| Iodine            | I <sub>2</sub>                     | 184,3              |
| Ammonia           | NH <sub>3</sub>                    | -33,3              |
| Phosphine         | PH <sub>3</sub>                    | -87,7              |

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

# Answers For Grade 11 Intermolecular Formal Task

**Lingsheng Yao**



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In a global consumed by screens and the ceaseless chatter of instant conversation, the melodic splendor and psychological symphony produced by the written term frequently diminish in to the backdrop, eclipsed by the constant sound and interruptions that permeate our lives. However, located within the pages of **Answeres For Grade 11 Intermolecular Formal Task** a charming fictional prize full of natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, this charming masterpiece conducts visitors on a psychological journey, skillfully unraveling the hidden melodies and profound impact resonating within each cautiously crafted phrase. Within the depths of the touching analysis, we can discover the book is key harmonies, analyze its enthralling publishing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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