# Chemical Formulas and Chemical Compounds

#### **SECTION 1**

SHORT ANSWER Answer the following questions in the space provided.

- In a Stock system name such as iron(III) sulfate, the Roman numeral tells us
  - (a) how many atoms of Fe are in one formula unit.
  - (b) how many sulfate ions can be attached to the iron atom.
  - (c) the charge on each Fe ion.
  - (d) the total positive charge of the formula unit.
- Changing a subscript in a correctly written chemical formula
  - (a) changes the number of moles represented by the formula.
  - (b) changes the charges on the other ions in the compound.
  - (c) changes the formula so that it no longer represents the compound it previously represented.
  - (d) has no effect on the formula.
- The explosive TNT has the molecular formula C<sub>2</sub>H<sub>2</sub>(NO<sub>2</sub>)<sub>3</sub>.
  - a. How many elements make up this compound?
  - b. How many oxygen atoms are present in one molecule of C<sub>2</sub>H<sub>3</sub>(NO<sub>2</sub>)<sub>3</sub>?
  - c. How many atoms in total are present in one molecule of C<sub>7</sub>H<sub>3</sub>(NO<sub>2</sub>)<sub>3</sub>?
  - d. How many atoms are present in a sample of 2.0 × 10<sup>23</sup> molecules of C<sub>7</sub>H<sub>5</sub>(NO<sub>2</sub>)<sub>3</sub>?
- 4. How many atoms are present in each of these formula units?
  - a. Ca(HCO<sub>3</sub>)<sub>2</sub>
    - b. C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>
    - c. Fe(ClO<sub>2</sub>)<sub>3</sub>
      - d. Fe(ClO<sub>3</sub>)<sub>2</sub>
- 5. \_\_\_\_\_\_ a. What is the formula for the compound dinitrogen pentoxide?
  - b. What is the Stock system name for the compound FeO?
  - c. What is the formula for sulfurous acid?
    - d. What is the name for the acid H<sub>3</sub>PO<sub>4</sub>?

# **Answers Review Chemical Formulas And Chemical Compounds**

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#### **Answers Review Chemical Formulas And Chemical Compounds:**

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