

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)
- At standard atmospheric conditions, all gases have the same average kinetic energy.
- At standard atmospheric conditions, all gases have the same average kinetic energy.

Model 1: Representation of Gases in Different Phases

The following figure shows the representation of particles in three different phases of matter.



Figure 1: Representation of particles in three different phases of matter.

At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)

Key Concepts

- At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)
- At standard atmospheric conditions, all gases have the same average kinetic energy, but are made of different molecules and are made up of different numbers of particles. (The kinetic molecular theory is based on the assumption that all particles have the same average kinetic energy.)

Figure 2: Representation of particles in three different phases of matter.

	Gas	Liquid	Crystalline Solid
Particle Motion and Kinetic Energy	Particles move in all directions, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.	Particles move in all directions, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.	Particles move in all directions, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.
Particle Arrangement and Kinetic Energy	Particles are arranged in a regular grid, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.	Particles are arranged in a regular grid, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.	Particles are arranged in a regular grid, with no net movement. They are moving in all directions, with no net movement. They are moving in all directions, with no net movement.

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